Subject	Monday	Tuesday	Wednesday	Thursday	Friday
ACCRS:	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.	se the periodic table as a model to predict the relative properties and trends (e.g., reactivity of metals; types of bonds formed, including ionic, covalent, and polar covalent; numbers of bonds formed; reactions with oxygen) of main group elements based on the patterns of valence electrons in atoms.
Before		Review dimensional analysis	Begin test	Unit 1 test grading and debrief	Data Set Quiz 1
During	Labor Day	Unit 1 Review	Unit 1 Test	Begin Unit 2/ Pass out Unit 2 outlines	Nearpod Lesson on section 2.1
After		Finish review	Finish test	Discuss first Section in Unit 2	Review
Desired Outcome		Students will review material covered for unit 1	Students will be tested on unit 1		Learn the basics of atoms and matter
Formative/ Summative		review	test	Class discussion	quiz